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Substituting  
and multiplying  
by the factor  
109 for the  
density unity  
kg/km<sup>3</sup>, the mass  
of the  
atmosphere is  
determined to  
be  $m = 5.092 \times 10^{18}$   
kg

Discussion

Performing the

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analysis with excel would yield exactly the same

results. EES

Solution for

final result:a=1

.2025166b=-0.101

67c=0.0022375r=6

377h=25m=4\*pi\*(a

\*r^2\*h+r\*(2\*a+b\*

r)\*h^2/2+(a+2\*b\*

r+c\*r^2)\*h^3/3+(

b+2\*c\*r)\*h^4/4+c



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$\rho \cdot h^5 / 5) \cdot 1E+9$

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pressure of refrigerant-134 on the volume of the tank is to be investigated as the initial pressure varies from 0.5 MPa to 1.5 MPa. The volume of the tank is to be plotted versus the initial pressure, and

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the results are to be discussed.  
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An Engineering approach. If all phases can be represented on a one-dimensional diagram (that is, a straight line representing composition), we are dealing with a two component system

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The results are:

$$\rho(z) = a + bz + cz^2 = 1.20252 - 0.101674z + 0.0022375z^2$$

for the unit of kg/m

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3, (or, ? (z) =  
(1.20252 -  
0.101674z +  
0.0022375z<sup>2</sup>  
) × 10<sup>9</sup> for the  
unit of kg/km<sup>3</sup>  
) where z is the  
vertical  
distance from  
the earth  
surface at sea  
level. At z = 7  
km, the equation  
would give ? =

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0.60 kg/m<sup>3</sup> .

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