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How many moles of hydrogen atoms are in 1 mole of each of the
following compounds? H_2O 2 CH_3OH 4 $(\text{NH}_4)_2\text{CO}_3$
8 2. Empirical and Molecular Formulas Give the empirical
formula for each compound: N_2H_2 NH BH_3 BH_3 H_2O
2 OH ...

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Stoichiometry. The atomic ratios in each compound are also the
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of products formed by a chemical reaction is called stoichiometry.

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10. methanol and oxygen gas 11. carbon dioxide and water 12. 160.10 g
13. 160.10 g 14. They are equal. 15. A mole ratio is a ratio between
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