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Conversation Skill

Practice 35 Gas Law

If the temperature of an ideal gas is raised from 100°C to 200°C , while the pressure remains constant, the volume

[A] remains the same
[B] doubles [C] goes to $1/2$ the original volume [D] increases by a factor of 100 [E] none of these

6. A

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4.37-g sample of a certain diatomic gas occupies a volume of 3.00 L at 1.00 atm and a temperature of 45 ° C.

Practice Test: Gas
Laws

Skill Practice 35

Name: _____ Date:

_____ Hour: _____

IMPORTANT:

whenever you use

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temperature, it must be in degree Kelvin (K), so remember the equation: $K = ^\circ C + 273$

1. a) convert $39^\circ C$ to K. b) convert 127 K to $^\circ C$. $312\text{ K} - 273 = 39^\circ C$

2. A gas has an initial volume of 2.75 L at a temperature of 285 K .

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Roosevelt High
School AP Chemistry

Answers
Gas Laws Practice

Gap-fill exercise. Fill in all the gaps, then press "Check" to check your answers. Use the "Hint" button to get a free letter if an answer is giving you trouble. You can also click on the "[?]" button to get a clue.

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Note that you will
lose points if you ask
for hints or clues!

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Answers to Gas Laws
Practice Problems. 1.

molar mass of $\text{Cl}_2 =$

$$2(35.45) = 70.90$$

$\text{g/mole} = 3.17 \text{ g/L. 2.}$

Molar volume is the
volume when $n =$

1.00 mole.

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Mixed Gas Laws
Worksheet 1) How
many moles of gas
occupy 98 L at a
pressure of 2.8
atmospheres and a
temperature of 292
K? 2) If 5.0 moles of

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0.2 and 3.0 moles of N_2O are placed in a 30.0 L tank at a temperature of 25 C, what will the pressure of the resulting mixture of gases be?

Mixed Gas Laws
Worksheet - Everett
Community College
Combined Gas Law.
The Combined Gas
Law combines

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Charles' Law,
Boyle ' s Law and Gay
Lussac ' s Law. The
Combined Gas Law
states that a gas'
(pressure \times
volume)/temperature
= constant. Example:
A gas at 110kPa at
30.0 ° C fills a flexible
container with an
initial volume of
2.00L.

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Gas Laws (video lessons, examples and solutions)

Gas laws - Wikipedia

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Giancoli Ch. 19 – p.
549-550, Problems
#35-45, 49-51 key;
Page 26/39

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Giancoli Ch. 20 – p.
577, Problems #1-8
key; ... This online
quiz is intended to
give you extra
practice with gas laws
problems. Select your
preference below and
click 'Start' to give it a
try! ... Ideal gas law
units to use (select at
least one for ideal gas
problems): Grams
Moles ...

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Gas Laws Test | Mr.
Carman's Blog

Mixed Extra Gas Law
Practice Problems

(Ideal Gas, Dalton ' s
Law of Partial

Pressures, Graham ' s
Law) 1. Dry ice is

carbon dioxide in the
solid state. 1.28

grams of dry ice is
placed in a 5.00 L

chamber that is

Download Free Skill Practice 35

maintained at 35.1°C.

What is the pressure
in the chamber after
all of the dry ice has
sublimed? $P = ?$

1.28 atm

1.1 atm

Extra Practice Mixed
Gas Law Problems
Answers

This worksheet and
quiz will let you
practice the following

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skills: Defining key concepts - ensure that you can accurately define main phrases, ... The Ideal Gas Law and the Gas Constant
8:03

Quiz & Worksheet -
Charles' Law |
Study.com
Extra Gas Laws
Practice Problems
Boyles ' , Charles ' ,

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Gas and Combined Gas

Laws 1) A sample of oxygen gas occupies a volume of 250. mL at a pressure of 740. torr. What volume will the gas occupy at a pressure of 800. torr if temperature is held constant? 2) A sample of nitrogen occupies a volume of 250 mL at 25 ° C. What volume will

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Gas Laws Extra
Practice eboard -
Garden City Public ...

GAS

STOICHIOMETRY

WORKSHEET Please
answer the following
on separate paper
using proper units
and showing ... 2.35 g

Fe(OH) 3 b. 1.76

Fe₂O₃ Problem 7:

4.40 g FeSO 4

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Problem 8: a. 0.5 mol
Al b. 0.75 mol H₂ c.
16.8 L H₂ Problem 9:
a. 312.5 L O₂ b. CO₂
= 200 L, H₂O = 225 L

GAS

STOICHIOMETRY

WORKSHEET -

PSD401

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QUESTION #10 A gas
at 2.5 atm and 25 ° C
expands to 750 mL
after being cooled to
0.0 ° C and
depressurized to 122

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kPa. What was the original volume of the gas? ANSWER #10

$P_1 = 2.5 \text{ atm}$ $T_1 = 25^\circ \text{C} = 298 \text{ K}$ $V_2 = 750 \text{ mL}$ $T_2 = 0.0^\circ \text{C} = 273 \text{ K}$ $P_2 = 122 \text{ kPa} = 1.20 \text{ atm}$ $V_1 = ? \text{ C.}$

Johannesson Gas
Laws Practice

Problems 1) Work out each problem on scratch paper.

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Gas Laws Practice

Problems - New

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Ideal gas equation.

Ideal gas equation. ...

Practice: Ideal gas
law. Practice:

Calculations using the
ideal gas equation.

This is the currently
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