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Describe how you would prepare 100 g of a solution that is 0.5% phenolphthalein by mass. Answer: Since the solute (phenolphthalein) is a solid, the solution is percent by mass. Mass percent means the number of grams of solute per 100 g of solution. mass percent = (mass of solute/mass of solution) x 100%.

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1. A 0.750 L aqueous solution contains 90.0 g of ethanol, C 2 H 5 OH. Calculate the molar concentration of the solution in mol · L<sup>-1</sup>.: Solution:

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