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Answer QUESTION 4 LAB REPORT

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Table 11.1: Specific heat. Specific  
heat of calorimeter: 0.22 cal/g.co  
Specific heat of water: Aluminum  
Samples Copper 65.39 m (g)  
190.39 47.89 m.

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ACTIVITIES CALORIMETRY*

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ACTIVITY 1: SP...  
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Calorimetry Lab Answers Correct Answer: A. Substance A A chemist mixes 500 g of lead at  $500^{\circ}\text{C}$  with 1,200 g of water at  $20^{\circ}\text{C}$ . She then mixes 500 g of copper at  $500^{\circ}\text{C}$  with 1,200 g of water at  $20^{\circ}\text{C}$ . The specific heat capacity

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of lead is  $0.1276 \text{ J/g}^\circ\text{C}$  and the specific heat capacity of copper is  $0.3845 \text{ J/g}^\circ\text{C}$ . Calorimetry Lab Flashcards | Quizlet  
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### Lab Answer Key

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reading for the week. You may want to print this document so that you can note observations as you work with the gizmo. Later, you can type your answers into the Word document and ...

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Gizmo™ to determine the specific heat capacities of various substances. 1. On the SIMULATION pane, select Copper. Use the slider to set its Mass to 200 g. Set the Water mass to 200 g. Check that the Water temp is set to 30.0 °C and the copper's is

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*CalorimetryLabSE.1.pdf - Name  
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Student Exploration: Calorimetry  
Lab Vocabulary: calorie,  
calorimeter, joule, specific heat  
capacity Prior Knowledge

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Questions (Do these BEFORE using the Gizmo.) 1. The Latin word calor means “heat,” and meter comes from the Greek word meaning “to measure.” What do you think a calorimeter does? Measures heat 2. Where have you heard the word calorie

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before? ? What do you think a  
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The results of Cheetos show a  
pattern of results being T1 (Least)  
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Highest and Higher than Least)

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burned. Sample #1 Sample #2  
Sample #3 Mass of Water 50 g 49  
g 50 g Initial Mass of Food Sample  
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Energy is required to heat up and cool down materials in the Calorimetry Lab Gizmo. In this Gizmo, various materials (copper, granite, lead, ice) are immersed in water inside an insulated container. The initial masses and temperatures of the materials

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